

NIKITINA, Ye.A.; TSVETKOV, N.A.

Potentiometric titration of β -luteophosphotungstic acid.
Zhur. neorg. khim. 8 no.10:2285-2289 0 '63. (MIRA 16:10)

1. 2-y Moskovskiy gosudarstvennyy meditsinskiy institut im. N.I.
Pirogova.
(Phosphotungstic acids) (Potentiometric analysis)

TELEGINA, Z.P.; SUBBOTA, M.I.; NIKITINA, Ye.A.

Characteristics of the distribution of hydrocarbon-oxidizing bacteria in the waters of the cross section of the Isbaskent oil and gas field. Mikrobiologiya 32 no.1:33-38 '63

(MIRA 12:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut yadernoy geofiziki i geokhimii.

NIKITINA, Ye. A.

[Heteropolymeric compounds. Heteropolysiloxanes. Moscow, Gos. nauchno-tekhn. izd-vo Khim. lit-ry, 1962. 421 p. (MIRA 17:6)]

DIKENSHTeyN, G.Kh.; KUTUZOVA, V.V.; MASHYKOV, K.K.; BABAYEV, A.G.;
POL'STER, L.A.; YUFEREV, R.F.; SHISHOVA, A.I.; BAREYEV,
R.A.; MAKAROVA, L.N.; MURADOV, K.; FYANOVSKAYA, I.A.;
SEMOV, V.N.; SIROTINA, Ye.A.; TURKINA, I.S.; FEL'MAN,
S.L.; KHON, A.V.; KUNITSKAYA, T.N.; GOLENKOVA, N.P.;
ROSHINA, V.M.; FARTUKOV, M.M.; SHCHUTSKAYA, Ye.K.;
ALTAYEVA, N.V.; BYKADOROV, V.A.; KOTOVA, M.S.; SMIRNOV,
L.M.; IBRAGIMOV, M.S.; KRAVCHENKO, M.F.; MARKOVA, L.P.;
RCZYIEVA, T.R.; UZAKOV, O.; SLAVIN, P.S.; NIKITINA, Ye.A.;
MILOGRADOVA, M.V.; BARTASHEVICH, O.V.; STAROBINETS, I.S.;
KARIMOV, A.K.

[Splicing of the wires of overhead power transmission lines]
Soedinenie provodov vozduzhnykh liniy elektroperedachi. Mo-
skva, Energiia, 1964. 69 p. (Biblioteka elektromontera,
no.132) (MIRA 17:9)

NIKITINA, Ye.A.; ROZENTOV, N.A.

Preparation of sodium 5-luteophosphotungstate. Zhur. obshch. khim. 10 no.12:2648-2652 1965. (MIR, 1965)

1. Vtoroy Moskovskiy meditsinskoy institut imeni Pirogova.

ЛЫТИНА, Ye F.

"The Environs of the Reservoir of the Kryukov Biological Station,
the Fauna, Its Biological Character, and the Possibility of Its Utilization."
Cand Biol Sci, Moscow Oblast Zoological Inst, Moscow, 1953. (Zhivotn., 1953,
Sep 54)

Sci: Sum 432, 29 Mar 55

1ST AND 2ND SECTORS										3RD AND 4TH SECTORS									
PROCESSES AND PROPERTIES INDEX																			
BC Microchemical examination of products of corrosion of aluminum and magnesium alloys. E. I. NIKITINA (Zaved. Lab., 1939, 8, 1058-1063).— The sample (20—12 mg.) is dissolved in HCl and then Mg and Al are determined by known hydroxyquinoline methods and Cu by pptn. with benzoxazine. For Zn, the sample is dissolved in 20% H₂SO₄, HNO₃ added, and the solution evaporated to dryness. The residue is dissolved in dil. AcOH, tartaric acid and 0.1 ml. of 0.5% CuSO₄·5H₂O are added, followed by 5 ml. of aq. K₂Hg(CNS)₄, and the ppt. of ZnCuHg₂(CNS)₄ (I) and CuHg(CNS)₂ (II) is collected and weighed; the amount of (II) corresponding with the Cu content of the solution is subtracted from the wt., and the Zn content is derived from the difference. Fe is determined by adding KI to the HCl solution of the sample, and titrating the liberated I. Mn is determined by dissolving the substance in 3—4 ml. of H₂SO₄ with 3—4 drops of HNO₃, and adding 0.5 ml. of 0.1N-AgNO₃ and 1 ml. of 10% (NH₄)₂S₂O₈. The coloration due to MnO₂ is compared with that of standard KMnO₄. R. T.										B-I-6									
ASB-51A METALLURGICAL LITERATURE CLASSIFICATION																			
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NOV 1961										NOV 1961									

CA

Determination of small quantities of sodium in metallic aluminum and silumin. R. I. Nikitina. *Zavodskaya Lab.* 6, 1947 80(1937). --An improved procedure for detg. Na as $\text{NaZn}(\text{UO}_2)_2(\text{AcO})_2$ of uniformly const. compn. in Al and silumin is described. Dissolve 0.5 g. of Al in 15-25 ml. HCl contg. 1-3 ml. of concd. HNO_3 , add excess NH_4OH , boil for 5 min., cool, dil. to 100 ml. and filter. Evap. an aliquot part (50 ml.) to dryness and ignite to

decomp. the NH_4 salts. Dissolve the residue in 15 ml. H_2O , filter and evap. to a 5-ml. vol. Stir the soln. with 5 ml. alc. and 10 ml. of Kolthoff's reagent (C. A. 21, 1773), let stand for 1 hr., then filter through a glass filtering crucible, continue the suction for 10-15 min., dry the ppt. in a desiccator for 1 hr. and weigh. To det. Na in silumin, dissolve 0.5 g. of sample in 25-30 ml. of aqua regia, expel N oxides and proceed as above. Sep. Na from the Cu by filtering the soln. from the insol. CuO formed in the decompn. of NH_4 salts. Chas. Blanc

66

7

Determination of small quantities of potassium in metallic aluminum and magnesium. B. I. Nikitina. *Zhurnal Khim. Lab.* 6, (1968) 1037. The gravimetric detn. of 0.1-1 mg. K in Al, Mg and their alloys as $K_2NaCo(NO_3)_6 \cdot H_2O$ is based on the insol. of the ppt. in $AcOH$ and the pptn. in the presence of excess Na salts and the formation of the monohydrate salt at 110° . The errors do not exceed 3% in the presence of other cations and 8% in the presence of more than 5% Cu and Fe. Evap. to dryness 0.3-0.5 g. of sample in HCl , dissolve the residue in a few drops of $AcOH$ and H_2O , filter from any SiO_2 , and introduce into the filtrate in the order given 1 ml. $AcOH$, 0.2 ml. alc., 5 ml. of concd. $NaCl$ and then quickly 5 ml. of 50% $NaNO_3$ and 5 ml. of 25% $Co(NO_3)_2$. Shake the soln. for 1 min., let stand for 24 hrs., filter through a Schott glass filter, No. 41, wash the ppt. 2-5 times with the prepd. soln., 10 ml. alc. and 10 ml. $AcOH$ in 100 ml. H_2O , then dry it at 110° for 1 hr. and weigh. Chas. Blanc

ASTM 3.5.4 METALLURGICAL LITERATURE CLASSIFICATION

M

PROCESSING AND PROPERTIES INDEX

*Microdetermination of Copper and Zinc in Aluminium and Magnesium Alloys. E. I. Nikitina (Zavod. Lab. (Works Lab.), 1938, 7, 102-106). C. Abs., 1938, 32, 1402. - (In Russian.) Cf. *Met. Abs.*, 1937, 4, 34. Cu is determined by Benedetti Pichler's method (*Met. Abs. J. Ind. Metals*), 1923, 30, 580) by electrolyzing the solution, containing 6-8 drops of HNO_3 and 3-4 drops of H_2SO_4 in 10 c.c. for each 10 mg. of the sample at 80°C. with Pt electrodes at 2-5-2-8 v. and 0-3-0-5 amp. The Montequi method (*Met. Abs. J. Ind. Metals*), 1920, 42, 532) of colorimetric detection of Zn as the violet salt $\text{CuZnHg}(\text{CNS})_2$ was made the basis of a microdetermination of Zn. Because of the minute amounts of Fe in the samples, the use of tartaric acid is unnecessary. Proceed with the decomposition and electrolysis of a 0-1 mg. sample, as above. Treat the solution (10-20 c.c.) with 3 drops of CH_3COOH , 0-1 c.c. of 0-5% CuSO_4 , and 10 c.c. of the reagent (20 grm. KCN and 10 grm. HgCl_2 in 500 c.c. of water), stir for 5 minutes, and, after standing for 1 hr., filter off the violet precipitate through a porous glass crucible. Wash the precipitate 5 times with water containing 0-5 c.c. of reagent per 100 c.c.; dry at 110°C. for 40 minutes, and weigh as $\text{CuZnHg}_2(\text{CNS})_4$. Deduct from the weight of the precipitate the $\text{CuHg}(\text{CNS})_2$ formed by the added Cu (factor 7-81) and calculate the Zn.—S. B. V.

E-2

137 AND 138 000485		140 AND 141 000485	
PROCESSING AND PROPERTIES INDEX			
C R		7	
New method for the determination of cadmium. R. I. Nikitina. <i>Zavodskaya Lab</i> 7, 400 (1968). The Mercuric test (C. A. 20, 2400) for the detection of Cd was developed into a method for detg. Cd in Al and Mg alloys. Unlike Mg, Cu, Zn, Mn and Al, Cd in the presence of KBr and bromine in dd. H₂SO₄ forms a white, cryst. ppt. of $(\text{H}_2\text{SO}_4)_2\text{CdBr}_2 \cdot 2\text{H}_2\text{O}$. To det. Cd in a Mg alloy, dissolve 0.5 g. of sample in 10-15 ml. of 20% H₂SO₄, evap. to 5 ml. and filter if any SO₂ is present. Introduce in quick succession 5 ml. of 1% bromine in 20% H₂SO₄ for a min. 5 mg. Cd) and 10 ml. of 10% KBr, stir vigorously until the pptn. is completed, let stand for 1 hr. and filter through a tared glass filtering crucible with suction. Wash the ppt. 5-6 times with the prepd. soln. (40 ml. of 1% bromine sulfate, 30 ml. of 10% KBr and 80 ml. H₂O) and then 8-10 times with alc. (EtOH 1:1), dry at 140-150° for 0.5 hr. and weigh. The caln. factor is 0.092. To det. Cd in Al alloys, dissolve 0.5 g. sample in a little HCl, evap. with 3 g. H₂SO₄ to fuming, dissolve the residue in H₂O, filter from SiO₂, evap. to 10-20 ml. and proceed as above. Chas. Blanc			
450-16.6 METALLURGICAL LITERATURE CLASSIFICATION			
13801 13801214		13801 13801214	
13801 13801214		13801 13801214	

1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									
PROCEDURES AND PROPERTIES, 1941 Determination of sodium in magnesium and its alloys. E. I. Nikitina. Zvezditskaya Lab. 8, No. 10-11, 1175-7 (1939); <i>Khim. Referat. Zhur.</i> 1940, No. 4, 53. —Na is pptd. as Na Zn uranyl acetate by Barber and Kuthoff's reagent (C. A. 22, 2000), the ppt. is dissolved in HNO ₃ , then reduced to III ⁺ by means of Al wire and the solution titrated with 0.1 N KMnO ₄ . This volumetric determination avoids errors arising from the formation of Na Mg uranyl acetate. The av. error of the detn. is 6%; individual variations reach 20%. The method is suitable for quantities of less than 0.1% Na. W. R. Heintz																			
ASB-51A METALLURGICAL LITERATURE CLASSIFICATION																			
1ST ORDER 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20										2ND ORDER 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40									
3RD ORDER 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60										4TH ORDER 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80									

PROCESSIES AND PROPERTIES INDEX																									
<i>04</i> Rapid method for determining zinc and copper in aluminum alloys. E. I. Nikitina. <i>Zavodskaya Lab.</i> 6, No. 12, 1314-15 (1939); <i>Khim. Referat. Zhur.</i> 1940, No. 5, 69. —Digest (0.2-0.5 g. of alloy with 10% H₂SO₄; all Cu remains in the residue. Add Na₂P₂O₇ to prevent pptd. of Fe (Al, Mg and Mn do not interfere) and ppt the Zn as Zn(Hg(CNS)₂). Filter, dry at 110°, and weigh. Cu can be detd. in the same sample by electrolysis after dissolving the residue insol. in H₂SO₄ in 6 N HNO₃. W R Heun <i>7</i>																									
ASAC-SLA METALLURGICAL LITERATURE CLASSIFICATION 140085																									

25

Spectroscopic and spectrodensographic methods of control in the process of dye manufacture. F. M. Shemyakin, E. I. Nikitina and K. I. Shklyareva. Bull. acad. sci. U. R. S. S., Ser. phys. 4, 120-1(1940). — Control in the production of the dianilide of purpurine and benzanthrone has been attempted by two different spectroscopic methods: (1) method of limiting dilution and (2) standardized absorption and spectrodensographic method. The results show a satisfactory accuracy. Rozalana Gamow

ASS. S. L. A. METALLURGICAL LITERATURE CLASSIFICATION

177
 Determination of Iron and Copper in Aluminium Alloys and of Iron in
 Stumias and Metallic Aluminium. E. I. Nikitina (Zavod. Lab. (Black) Lab.),
 1940, 8, 225-227; C. Abe., 1940, 34, 6188. [In Russian.] The method
 employed in works' laboratories for determining Fe in Al alloys by dissolving
 in hot alkali hydroxide gives low results; some of the Fe dissolves in the hot
 NaOH solution. In an improved method, the sample is dissolved in 10%

H₂SO₄, the undissolved Cu precipitate is filtered off, and the filtrate mixed
 with 20 ml. of HCl (1:1) after which 0.1 gram. of metallic Al is added. The
 solution is cooled under water for 15 minutes, 5 ml. of 7% H₃PO₄ solution
 is added, and the titration is carried out with 0.5N K₂Cr₂O₇ in the presence
 of diphenylamine to a violet blue colour. To determine Cu, the Cu precipi-
 tate is dissolved in HNO₃ (1:1), 1 ml. concentrated H₂SO₄ is added and the
 solution is electrolyzed.

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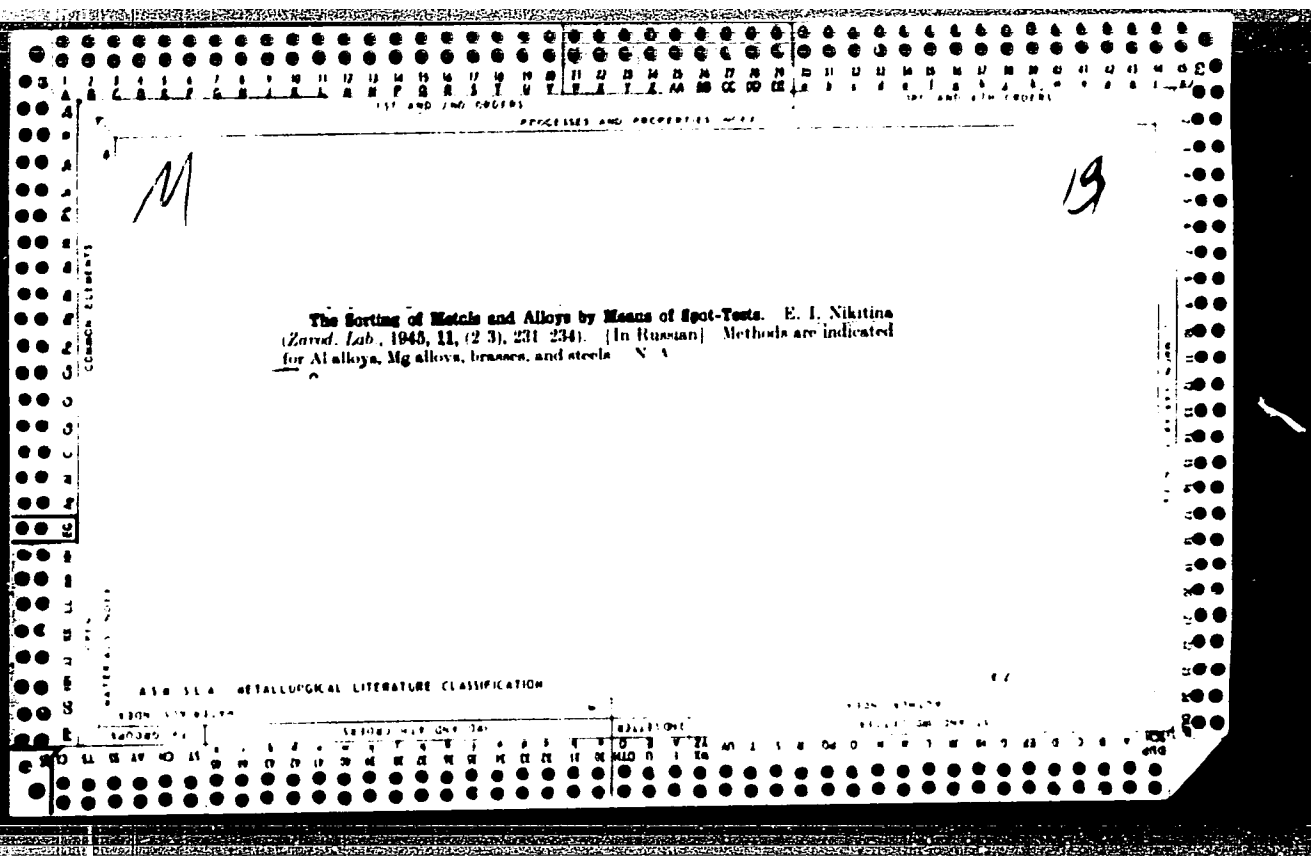
m

***Determination of Iron in Light Alloys by Means of Sulphosalicylic Acid.**
F. L. Nikulina (*Zavod. Lab.*, 1940, 9, (30), 020-030). In Russian. N.
recommends the introduction of NH_4Cl to prevent precipitation of $\text{Mg}(\text{OH})_2$.

ASAC METALLURGICAL LITERATURE CLASSIFICATION

E.I. NIKITINA

*A Rapid Method for the Determination of Zinc in Aluminum Alloys by
the Simultaneous Determination of Copper. *J. Anal. Chem.* 1962, 37, 1952-1954. Translated from *Zh. Anal. Khim.* 1962, 17, 2200-2201.



1ST AND 2ND ORDER																									
PROCESSES AND PROPERTIES																									
CA A rapid method for the determination of vanadium in steel by means of dimethyl <i>p</i>-phenylenediamine. F. J. Nibbeling. <i>Analyst</i> 101, 1041 (1976). To 0.1 g of the sample in 20 ml of HNO₃ (1:1) and 1 ml of HCl (1:1), evap. to 5 ml., add 100 ml. of water, 1 g. of NH₄F (the soln. becomes turbid, the reaction is weakly alk., and the soln. is colorless), 20 drops of H₂PO₄, an addnl. 0.5 g. of NH₄F, and 2 ml. of 0.1% aq. dimethyl <i>p</i>-phenylenediamine (a pink color appears in the presence of V), and titrate the soln. immediately with 0.01 N NaOH, adding it slowly with const. mixing until the soln. is fully decolorized. To verify the completeness of reduction by NaOH, add an addnl. 1 ml. of dimethyl <i>p</i>-phenylenediamine and titrate until no color appears after each addn. of the reagent. Good results were obtained. The detectable quantities of V were from 0.01% to 2 mg. Three references. W. R. Hunt.																									
010 55 A METALLURGICAL LITERATURE CLASSIFICATION ESSEN: 170103100 170103100																									

1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									
PROCESSES AND PROPERTIES INDEX																			
S										21									
SPECTROGRAPHIC DETERMINATION OF ARSENIC IN STEEL. E.I. Nikitina. (Zavodskaya Laboratoriya, 1947, vol 13, p 886; Chemical Abstracts, 1948, vol 42, Nov. 20, col. 8899). The method was developed to determine arsenic in steel made from iron ores containing arsenic. A large autocollimation spectrograph of the Hilger type was used. The source of excitation was an A.C. arc. The most satisfactory line for analysis, based on intensity and distribution, is the line As I 2349.84A. which is sufficiently sensitive to arsenic concentrations encountered (0.085%, 0.13%, and 0.36%). The neighbouring line Fe I 2550.39A. was used for comparison. The lines were photometered by the method of photometric interpolation. A pointed carbon rod was used as the second electrode; the use of copper and iron rods as constant electrodes gave less satisfactory results. The width of slit in the spectrograph was 0.02 mm., distance between electrodes																			
ASS. SLA METALLURGICAL LITERATURE CLASSIFICATION																			
1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									

was 2 mm., current strength was 10 amp., and exposure time was one minute. A medium-type model spectrograph can be used for these analyses.

9-4

M

*Spot Analysis in Sorting Stellite, Beryllium, Steels, Chromanils, and Bronzes.
E. I. Nikitina (Zavod. Lab., 1947, 12, 923-925; C. Abstr., 1950, 44, 3404). [In Russian]. A number of spot tests carried out on the alloys are described.
Cr-Ni Stellites are identified by the reaction of Ni with dimethylglyoxime.
Cr-Fe base Stellite is identified with $K_2Fe(CN)_6$. Be steels are tested with
quinizarin. Chromanils are identified by testing for Mn with $(NH_4)_2S_2O_8$.
Bronzes are identified similarly by testing for one of the constituents. For
most of these tests a drop of acid is placed on a cleaned spot on the metal,
the dissolved matter is sucked up on filter paper, and the appropriate reagent
is applied to the spot on the filter paper. In some cases the drop of acid and
its content are transferred to a small crucible and the test is finished there.

NIKITINA, YE. I.

PA 62T74

USSR/Metals

Mar 1948

Zinc - Determination
Alluminum Alloys

"Microchemical Determination of Small Amounts of Zinc by Dithizone (Diphenylthiocarbazone) in Aluminum Alloys," Ye. I. Nikitina, All-Union Inst Aviation Materials, 3 pp

"Zavod Lab" Vol XIV, No 3

Method permits determination of zinc content from several γ to one or two tenths of γ . Possible to determine ten hundredths percent of zinc in spite of the large presence of aluminum. Method is simple, however, much preparation is necessary to purify the reagents due to the sensitivity of dithizone.

62T74

NIKITINA, YE. I.

PA 11/1971

USSR/Chemistry - Analysis, Quantitative Apr 48
Chemistry - Iron Alloys, Grading of

"The Use of the Drop Method for Approximate
Quantitative Analysis in the Grading of Alloys,"
Ye. I. Nikitina, All-Union Inst of Aviation
Materials, 5 pp

"Zavod Lab" Vol XIV, No 4

Describes how method is applied to determination of
copper, silicon and manganese. Editor notes, how-
ever, that total error (sum of colorimetric and
solution errors) may amount to 30%.

4/49718

The Colorimetric Determination of Small Quantities of Antimony in Copper and Tin Bronzes. E. I. Nikitina (Zarod. Lab., 1948, 14, (8), 933-935). (In Russian). The method is based on the colorimetric estimation of the yellow complex KSbI_4 in sulphuric acid solution in the presence of ascorbic acid. For Cu the method is as follows: 1-2 g. of sample is dissolved in HNO_3 (density 1.4); the solution is evaporated to a small volume, made up to 30 ml. with water, 2 ml. of 10% MnSO_4 solution and 20 drops of 4% KMnO_4 solution added, and the solution boiled 20-30 min. in a covered beaker. The precipitate is filtered off and freed from Cu by washing well with hot water. The precipitate is dissolved on the filter by hot HCl containing a few drops of H_2O_2 , H_2SO_4 is added, and the volume of solution reduced to half by boiling. NH_3 is added in excess, a further 10 drops of KMnO_4 added, and the solution boiled for 20 min. The precipitate is again filtered off, carefully washed with hot water containing NH_3 , and re-dissolved in hot HCl containing H_2O_2 . To the 5 ml. of 16% H_2SO_4 is added and the solution evaporated to dryness. To the residue a further 5 ml. of 16% H_2SO_4 is added, and should the solution have a yellow colour, 5 ml. HNO_3 is added and the solution boiled to colourless. The procedure may need to be repeated until the solution is colourless. The solution is diluted with 16% H_2SO_4 and transferred to a cylinder for colorimetry, washing out the beaker with more 16% H_2SO_4 . The volume of liquid should not be more than 15-20 ml., and to it is added 4-5 ml. of KI solution containing ascorbic acid (12 g. KI and 1 g. ascorbic acid in 100 ml. water). Into an identical cylinder the same quantity of 16% H_2SO_4 is run and 4-5 ml. of KI solution containing ascorbic acid. A standard solution of Sb in ml. of KI solution containing ascorbic acid. The colour is identical with the specimen under test. The procedure needs some modification for the determination of Sb in tin bronzes. N. B. V.

ASB SLA METALLURGICAL LITERATURE CLASSIFICATION

62/49T82

UssR/Metals
Carbon
Test Techniques

Jul 49

"Microchemical Determination of Carbon in
Regular and Alloyed Steels," A. I. Glazov,
Ye. I. Nikitina, All-Union Inst of Avn Mate-
rials, 2 pp

"Zavod Lab" No 7

Aim of authors was to develop method not re-
quiring complex apparatus. A Wurtz apparatus
was reconstructed somewhat: gas-measuring
burette was replaced by a micro-burette cap-
able of measuring volumes to 0.01 ml, suitable
62/49T82

UssR/Metals (Contd)

Jul 49

For working with samples 0.2 g or less. Tabu-
lated data shows absolute error and deviation
of parallel tests for miscellaneous steels.
Accuracy of method was ± 0.01 for 0.15% and
higher carbon content and ± 0.005 for 0.1% or
lower.

62/49T82

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Microchemical Determination of Carbon in Ordinary and Alloy Steels. A. I. Olazova and E. I. Nikitina. (Zavodskaya Laboratoriya, 1949, vol. 15, July, pp. 847-848). (In Russian). A modified Wirtz apparatus with a microburette reading to 0.01 ml. is said to have been satisfactorily used for the gravimetric determination of carbon in steels, the accuracy being: $\pm 0.01\%$ and $\pm 0.005\%$ for carbon contents of over 0.15%, and 0.1% or less, respectively. —a. n.

ASTM-SLA METALLURGICAL LITERATURE CLASSIFICATION

COMMON ELEMENTS

PROCESSING AND PROPERTIES INDEX

1ST AND 2ND EDITIONS

3RD AND 4TH EDITIONS

COMMON ELEMENTS

PROCESSING AND PROPERTIES INDEX

1ST AND 2ND EDITIONS

3RD AND 4TH EDITIONS

NIKITINA, Yekaterina Ivanovna; GRACHEV, K.Ya., redaktor; LUR'YE, M.S.,
tekhnicheskiiy redaktor

[Accelerated semimicrochemical methods for analyzing metals and
alloys] Uskorennye polumikrokhimicheskie metody analiza metallov
i splavov. Moskva, Gos. nauchno-tekhn. izd-vo khim. lit-ry, 1956.
306 p. (MLRA 9:9)

(Metals--Analysis)

NIKITINA, E.I.

V. Photocolorimetric determination of phosphorus in the presence of tungsten, titanium, and niobium. ²⁷
~~Nikitina, Zvezdskaya Lab., 22, 1027-30 (1959).~~ A photo-
 colorimetric method for the detn. of small amts. of P in

Alloys when W, Ti, and Nb are present was based on the measurement of the blue color intensity of the reduced phosphomolybdenum compd. with thiourea and CuSO_4 . The alloy was dissolved in aqua regia, some HF added, stirred for 1 min., 2 g. H_2BO_3 was added, dild. to 100 ml., and a 10-ml. aliquot portion was filtered through a double filter. The sample was neutralized with NH_3 , any $\text{Fe}(\text{OH})_3$ formed redissolved with a few drops of 6N H_2SO_4 , 4-5 ml. 6N H_2SO_4 , 1 ml. 1% CuSO_4 , 10 ml. thiourea (10% soln.), and finally 4 ml. of a 5% $(\text{NH}_4)_2\text{MoO}_4$ soln. added, and the color intensity detd. in a photocolorimeter. W, Ti, and Nb present must be previously converted into complexes, Ti with HF (with the addn. of H_2BO_3 to take care of the excess of HF). When W is present, the aqua regia soln. must be evaporated to dryness, the dry residue redissolved in 15 ml. HNO_3 and 15 drops HF added. In the presence of W and Nb more HF must be added and the soln. boiled for 30 min. for complete hydrolysis of WO_3 . When W and Ti are present, the evapn. to dryness must be omitted, and good results are obtained by using H_2O_2 with HF. With all the 3 potentially interfering metals present, the H_2O_2 must also be used, and the phosphomolybdate reduced as given above, taking care of the proper amt. of acid to give a pH of 0.8-0.9, and waiting 20 min. for the color to develop.

W. M. Sternberg

MT

NIKITIN, E. I.

27 27 18
Determination of tantalum and niobium in steel. E. I.
Nikitin. U.S.S.R. 116,600, June 26, 1957. A mixt. of
Ta and Nb is detd. photocolorimetrically by using ammonium
soln. as the reagent. M. S. 11

5
4E2C

16

NIKITINA, E. I.

Distr: 4E2c/4E4j

27 27 4
2
Photocolorimetric determination of iron in titanium. E. I.
Nikitina and N. T. Slinko. U.S.S.R. 107,911, Oct. 7, 1967.
Fe is detd. without sepg. the Ti. The detn. is made
with the aid of sulfosalicylic acid in an ammoniacal medium
so that the Ti is fixed in a colorless complex. M. Horach 11
PM

AUTHOR Malyarov K.M., Nikitina Ye.I., 32-7-48/49
 TITLE The Accelerated Semimicrochemical Methods of Analyzing Metals and Alloys, by Nikitina, Ye. I.
 (Uskorennyye polumikrokhimicheskiye metody analiza metallov i spavov.. Ye. I. Nikitina)
 PERIODICAL Zavodskaya Laboratoriia, 1957, Vol 23, Nr 7, pp 887-887 (U.S.S.R.)
 ABSTRACT The book consists of 4 parts. The first part describes the working processes, and the manner in which samples are taken for quantitative microanalyses, and possible errors committed are described. The remaining three parts of the book contain a detailed description of suitable preparations for qualitative microdeterminations and quantitative semimicrodeterminations. All parts of the book are arranged according to one and the same plan: qualitative analysis is carried out by microchemical and quantitative analysis by semimicrochemical methods. Some of the material is the result of the research work carried out by the authoress, as e.g. the qualitative method of analyzing aluminum- and magnesium alloys, testing of metal coatings, etc. As a qualitative analysis the microcrystalloscopic method is preferred. This method is illustrated by a micropicture of the crystals. What is missing in the book is, however, above all, the titration description by trilon "B". The book is recommended for the analysis of special alloys.
 ASSOCIATION Moskovskiy gosudavstvennyy universitet
 AVAILABLE Library of Congress.
 Card 1/1

AUTHOR: Nikitina, Ye. I. 75-1-11/26

TITLE: The Photometric Determination of Tantalum in Titanium and Its Alloys by Means of Arsenazo as a Reagent
(Fotometricheskoye opredeleniye tantala v titane i yego splavakh pri pomoshchi reagenta arsenazo)

PERIODICAL: Zhurnal Analiticheskoy Khimii, 1958, Vol 13, Nr 1, p. 72-78 (USSR)

ABSTRACT: In the determination of tantalum and niobium in different materials their separation from titanium proves to be the most difficult task as titanium reacts with the same reagents as tantalum and niobium. The present paper describes a photo-colorimetric method of determination for tantalum which is applicable without a previous separation of titanium. The reagent arsenazo, or benzen-2-arsonic acid-(1-azo-7>1,8-dihydroxy-naphthalene-3-6-disulfonic acid, serves for the photometric determination of rare earth metals, tetravalent vanadium, zirconium, aluminum and niobium (refs. 3-6). Neutral and acid aqueous solutions of arsenazo have a pink color. In the presence of tantalum as well as of niobium the color becomes

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The Photometric Determination of Tantalum in Titanium and
Its Alloys by Means of Arsenazo as a Reagent

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violet. This color reaction only takes place in highly acid solutions with p_H -values of 2 or less. At p_H 1-2 the intensity of the color in an acetic or hydrochloric solution is highest. The color in a hydrochloric solution lasts several days, is well reproducible and therefore suitable for photometric determination. At a tantalum content of from 0,03 to 1 mg the coloring of the tantalum complex with arsenazo follows Beer's law. The sensitivity of the reaction amounts to 0,3 γ per ml and is therefore much higher than in the reaction of tantalum with pyrogallol. The complex is resistant to heating to 100°C. The corresponding measurements were performed on a photocolormeter of the type Φ JK-H with the light filter no. 6. Complex-forming ions can influence the reaction of tantalum with arsenazo. Oxalates and fluorides form very stable complexes with tantalum at room temperature and therefore prevent the reaction with arsenazo. The complexes of tantalum with tartrate and citrate ions in acid solutions are not so stable and do not hinder determination. Determination can therefore be performed in tartrate- and citrate-containing solutions. The presence of free tartaric acid, however, lowers the intensity of the coloring of the tantalum complex,

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The Photometric Determination of Tantalum in Titanium and
Its Alloys by Means of Arsenazo as a Reagent

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and with a large excess of tartaric acid complete decolorization may occur. A uniform concentration of tartaric acid must therefore be maintained in the quantitative determination of tantalum with arsenazo. At room temperature in an acid solution titanium forms a violet coloring with arsenazo. On increasing the acidity to a pH of 0,5 the complex is, however, completely destroyed, especially rapidly on heating of the solution to 100°C. Therefore titanium does not disturb the determination of tantalum at a pH < 0,5. The presence of titanium, however, causes a slight decrease in the intensity of the color. For this reason a calibration curve has to be set up. In the presence of 5 to 10 mg Ti the intensity of the coloring of the tantalum complex follows Beer's law in an interval of from 0,05 to 0,8 mg tantalum. With a titanium content > 20 mg work has to be done in a still stronger acid solution. According to this method a simple and rapid photocolometric method for the determination of from 1 to 15 % tantalum in titanium alloys was worked out. The chief quantity of titanium does on that occasion not disturb tantalum determination. An aluminum content of up to 10 % and an iron

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The Photometric Determination of Tantalum in Titanium and
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content of up to 1 % do not disturb determination. A higher iron content reduces the intensity of the color and may even lead to decolorization. Alkali and earth-alkali metals do not disturb determination. The presence of less than 1 mg zirconium in the solution which is photometered does not disturb either. The corresponding complex of niobium with arsenazo has the same color as the tantalum complex and has also the same properties of formation. When, therefore, niobium is present, the sum of niobium and tantalum is obtained by the determination with arsenazo. Consequently niobium disturbs the determination of tantalum alone. There are 5 figures, 6 tables, and 6 references, all of which are Slavic.

SUBMITTED: March 21, 1957

AVAILABLE: Library of Congress

1. Tantalum - Determination
2. Arsenazo - Reagent
3. Niobium - Determination
4. Titanium - Chemical reactions

Card 4/4

AUTHOR: Nikitina, Ye.I.

32-24-4-5/67

TITLE: The Photocolorimetric Determination of Silicon in Copper Alloys
(Fotokolorimetriceskoye opredeleniye kremniya v mednykh splavakh)

PERIODICAL: Zarodskaya Laboratoriya, 1958, Vol. 24, Nr 4, pp. 398-402 (USSR)

ABSTRACT: Ye.N. Yegorova (Ref 3) carried out investigations of the transformation of silicic acid ($\alpha \rightleftharpoons \beta \rightleftharpoons \gamma$) under the effect of ammonium nitrate and ammonium sulfate as well as of various temperatures and concentrations in the case of a content of 800 mg/l SiO_2 . Basing upon this work the present determinations, the experimental part of which was carried out in co-operation with N.T. Slinko, were worked out. The results obtained are shown in form of diagrams. It was found that for the colorimetric determination of silicon an acid concentration of 0.02 - 2n is the optimum; decomposition of the sample should be carried out at room temperature with nitric acid, i.e. boiling or long heating should be avoided. Lead-, antimony-, iron- and aluminum salts do not disturb this determination (if the concentration of SiO_2 does not exceed

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The Photocolorimetric Determination of Silicon
in Copper Alloys

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1 mg/100 ml), whereas even small quantities of manganese and nickel salts cause polymerization of silicic acid. In order to prevent this, ammonium persulfate was used in investigation of various types of bronze. Two processes of analysis for bronze and antimony-phosphorus bronze are given; the former can be investigated in the case of a content of 0.1 to 1.0% silicon with an accuracy of $\pm 0.02\%$, and the latter at 0.002 to 0.1% silicon with an accuracy of $\pm 0.001\%$. If the latter contains more than 0.1% silicon, work is carried out according to the first process of analysis, but there is no addition of ammonium persulfate. There are 4 figures, 3 tables, and 6 references, 4 of which are Soviet.

1. Copper alloys--Analysis 2. Silicon--Determination 3. Nitric acid--Metallurgical effects 4. Colorimetry--Applications

Card 2/2

NIKITINA, Ye. I.

18(6) PHASE I BOOK EXPLOITATION SOV/3199
Amdenya nauk SSSR. Institut obshchey i neorganicheskoy khimii
im. N. S. Kurnakova

Analiz blagoderzhnykh metallov (Analysis of Noble Metals) Moscow,
1959. 193 p. Irata slip inserted. 2,700 copies printed.

Resp. Ed.: N. K. Pribitayn. USSR Academy of Sciences. Corre-
sponding Member and O. Ye. Zvyagintsev, Doctor of Chemical
Sciences; Eds. of Publishing Houses: T. O. Levi, and D. N.
Trifonov; Tech. Ed.: I. M. Guseva.

PURPOSE: This collection of articles is for scientists engaged
in the study and analysis of the noble metals.

COVERAGE: This is a collection of articles on the analysis of the
noble metals. It includes studies carried out by the Institute
of General and Inorganic Chemistry in Moscow (AN SSSR),
as well as studies presented at the Third and Fourth Conference
on Noble Metals held in 1954 and 1957, respectively. The
studies and reports describe new organic reagents for gravi-
metric determination of platinum metals, and physicochemical
methods of analysis (spectrophotometric, polarographic and
potentiometric). Special attention is given to spectral
analysis for the determination of admixtures in alloys of
platinum metals, silver, and gold, as well as in refined noble
metals. The collection also includes analytical methods, tables
and charts for materials containing metals of the platinum
group, as well as a review of the literature on the analysis
of platinum metals published in the last five years. No
personalties are mentioned. References follow each chapter.

Pribitayn, N. K., K. A. Gladyshevskaya and L. M. Ryabkova.
Use of the Ion Exchange Method in the Analysis of Platinum
Metals. Report 2. Separation of Rhodium from Iridium 103

Anisimov, S. M., Ye. I. Nikitina and V. M. Alyanchikova.
Methods of Preparing Poor Thermoplastic Solutions and Obtaining
From Them Cemented Substances for the Determination of
Platinum Metals by Spectral Analysis 115

Khrapay, V. P. Spectral Method for the Determination of
Platinum, Palladium, and Tellurium in Silver-gold Alloys 128

Pribitayn, N. K. and A. D. Ostikova. Spectral Method of
Analysis for Refined Iridium and Ruthenium 133

Kuranov, A. A., N. P. Rukhs and M. M. Skviridova. Spectral
Determination of Mixtures in Gold, Silver and Alloys 139

Kuranov, A. A. Spectral Analysis of Platinum Alloys Con-
taining Three Components 143

Adakhovskiy, A. P. and V. M. Karbolin. Determining the
Chemical Composition of Binary Alloys by the Thermoelectro-
motive Force 145

Arlov, V. B. Effect of Complexation and of the Acid-
Alkali Balance in the Medium on the Potential of the
Au^{III}/Au⁰, Au^I/Au⁰, Au^{III}/Au^I, and Ag^I/Ag⁰ Systems 150

Arlov, V. B. and Y. V. Kosova. Chromatometric Determination
of Gold 156

Anisimov, S. M., V. M. El'yankov and V. P. Tsvetkov.
Spectrometric Method for the Determination of Silver in
Silver and Lead Alloys Containing Platinum Metals 163

Yufa, T. P. and M. A. Chentseva. Dissolving Platinum
Metals and Their Alloys with the Aid of an Alternating
Current 176

Chentseva, M. A., T. P. Yufa and Y. G. Levilash. New
Method for the Analysis of Palladium-silver Alloys 181

Rushnikov, M. S. and K. S. Sheina. Methods of Testing
Palladium Alloys and Their Products on a Touchstone
and by Chemical Means 184

NIKITINA, Ye. I.

5(2)

PHASE I BOOK EXPLOITATION

SOV/3224

Mukhina, Zinaida Stepanovna, Yekaterina Ivanovna Nikitina, Lidiya Mitrofanovna Budanova, Raisa Samuilovna Volodarskaya, Lyudmila Yakovlevna Polyak, and Anna Aleksandrovna Tikhonova

Metody analiza metallov i splavov (Methods of Analysis of Metals and Alloys) Moscow, Oborongiz, 1959. 527 p. Errata slip inserted. 8,050 copies printed.

Ed. of Publishing House: T. M. Kunyavskaya; Tech. Ed.: V. P. Rozhin.

PURPOSE: This book is intended for laboratory technicians of plants and may also be of use to personnel of chemical and analytic laboratories of scientific institutions and schools of higher education.

COVERAGE: The book reviews various methods of analyzing steel, cast iron, complex iron, chromium-, nickel- and cobalt-base alloys. It also reviews methods of determining the content of elements in aluminum, magnesium and copper alloys as well as in various binary alloys. Principles of physical and chemical analysis for

~~Card 1/1*~~

5(2), 5(3)

SCV/75-14-4-8/30

AUTHOR: Nikitina, Ye. I.

TITLE: Photometric Determination of Titanium in Titanium Borides
by Means of Arsenazo

PERIODICAL: Zhurnal analiticheskoy khimii, 1969, Vol. 11, Nr. 4,
pp 431 - 433 (USSR)

ABSTRACT: Arsenazo, otherwise benzene-2-azobenzic acid<1-azo-2>1,8-di-
hydroxynaphthalene-3,6-disulfonic acid, is pink-colored in
aqueous neutral and acid solutions. In the presence of ti-
tanium, this coloring changes at pH 2-3 to violet. With in-
creasing acidity of the solution this coloring is weakened
and disappears entirely at pH 3-0.9 (Ref 4). At pH 2.5-3,
the colored complex compound of titanium with arsenazo is
most constant. In this pH-interval, the colored solutions
with titanium contents of from 0.01-0.25 mg obey the Beer
law (Fig 2). The coloring of the complex is well reproducible
and very suitable for the quantitative photometric determina-
tion of titanium. At room temperature, the coloring develops
immediately after adding the reagent to the weakly acid solu-
tion of a titanium salt. The coloring lasts some days, and is

Card 1/3

Photometric Determination of Titanium in Titanium
Borides by Means of Arsenazo

SOV/75-14-4-8/30

also not changed by heating. The reaction of titanium with arsenazo is much more sensitive than that with hydrogen peroxide. The sensitivity amounts to 0.05% of titanium per ml. The complex forms at pH 2-3 in hydrochloric and sulfuric solution. When determining titanium arsenazo, it is expedient to make the photometric determination of the mixed coloring from the violet colored complex and the color of the excess reagent. This mixed coloring is obtained by adding 10 ml of a 0.05 per cent solution of the reagent, at titanium contents of from 0.01-0.25 mg/100 ml of the solution. The absorption maximum of the complex at pH 2.5-3 is found at 580 m μ (Fig 3). In this spectral range, the pure reagent possesses a minimum of absorption. The results of 7 determinations of titanium by this method are shown in table 1. In the presence of oxalates and fluorides, the colored complex of titanium does not develop. Tartaric and citric acid slightly reduce the color intensity in sulfuric solution, but do not affect the sensitivity of the reaction. The influence of these masking compounds is shown in table 2. Alkali- and alkaline-earth metals do not disturb the determination. Trivalent iron (up to 0.03 mg),

Card 2/3

5(2)

AUTHOR:

Nikitina, Ye. I.

SOV/32-25-2-6/78

TITLE:

Complexometric Determination of Zirconium in Borides and Nitride (Kompleksometricheskoye opredeleniye tsirkoniya v boridakh i nitride)

PERIODICAL:

Zavodskaya Laboratoriya, 1959, Vol 25, Nr 2, p 142 (USSR)

ABSTRACT:

In order to determine larger quantities of zirconium in its borides and nitride (most of which are soluble in sulfuric acid) it is suggested to titrate the material with trilon and to carry out a backtitration of the complexon surplus with nitric bismuth salts, thiourea being used as indicator (Ref 1) at a pH of 2. In this way up to 90 % zirconium can be determined with an accuracy of ± 0.4 %. The analysis can be carried out in the presence of sulfuric acid and other acids as well as the majority of bivalent cations (Table). Zirconium forms a number of different complex compounds with trilon (Ref 2). It was also observed that white insoluble compounds form which are dissolved when being heated. The analysis process is described. There are 1 table and 2 references.

Card 1/1

NIKITINA, Ye.I.

Determining antimony impurities in pure chromium and its alloys.
Trudy Kom. anal. khim. 12:311-313 '60. (MIRA 13:8)
(Chromium--Analysis) (Antimony)

PHASE I BOOK EXPLOITATION

SOV/6035

Mukhina, Zinaida Stepanovna, and Yekaterina Ivanovna Nikitina

Uskorennyye metody analiza titana i yego splavov (Accelerated Methods of Analyzing Titanium and Titanium Alloys) Moscow, Oborongiz, 1961. 121 p. Errata slip inserted. 5050 copies printed.

Ed.: T. M. Kunyavskaya; Tech. Ed.: L. A. Garnukhina; Managing Ed.: A. S. Zaymovskaya, Engineer.

PURPOSE: This book is intended for technical personnel of scientific research institutes and analytical laboratories of metallurgical and machinery plants. It can also be used in schools of higher education.

COVERAGE: The book describes chemical and (mainly) physicochemical methods of determining all the components of industrially produced titanium alloys. Optimum methods, selected from several cited, for determining each of the com-

Card 1/02

ZAKHAROVA, Galina Vasil'yevna, kand. tekhn. nauk; POPOV, Ivan Alekseyevich, kand. tekhn. nauk; ZHOROVA, Lilianna Pavlovna; FEDIN, Boris Vladimirovich; Prinsipali uchastiye: MUKHINA, Z.S., zasl. deyatel' nauki i tekhn. RSFSR; POPOVA, I.A., zasl. deyatel' nauki i tekhn. RSFSR; YEGOROVA, N.D., zasl. deyatel' nauki i tekhn. RSFSR; NIKITINA, Ye.I., zasl. deyatel' nauki i tekhn. RSFSR; ZHEMCHUZHAYAY, Ye.A., zasl. deyatel' nauki i tekhn. RSFSR; ZHABINA, V.A.; SAVITSKIY, Ye.M., red.; STROYEV, A.S., red.; ARKHANGEL'SKAYA, M.S., red. izd-va; KARASEV, A.I., tekhn. red.

[Niobium and its alloys] Niobii i ego splavy. By G.V.Zakharova i dr. Moskva, Gos. nauchno-tekhn. izd-vo lit-ry po chernoi i tsvetnoi metallurgii, 1961. 368 p. (MIRA 14:12)
(Niobium)

25353

S7032/61/027/006/004/018

B124/B203

55300

AUTHOR:

Nikitina, Ye. I.

TITLE:

Photocolorimetric determination of niobium in a titanium alloy and an aluminum alloy with the reagent "Arsenazo"

PERIODICAL:

Zavodskaya laboratoriya, v. 27, no. 6, 1961, 663 - 666

TEXT: Niobium gives with Arsenazo a violet, soluble compound at pH = 2. The violet color of the Nb-Arsenazo complex is stable for some days, well reproducible, and is formed immediately on addition of the reagents. The mixed color (Arsenazo is pink and the niobium complex violet) is measured in the photocolormeter at a wavelength of 580 mμ with a yellow-orange light filter. The absorption maximum of the niobium complex in hydrochloric acid solutions and the absorption minimum of solutions of pure Arsenazo lie in this spectral range. The author studied the effect of tartaric acid on photocolormetric niobium determination; results given in Table 1 show that the reaction with Arsenazo proceeds quantitatively at a content of less than 0.04 g of tartaric acid in the colorimetric solution. With an increase in the tartaric acid content to 0.1-0.2 g/100 ml, the optical density of the

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Photocolorimetric determination...

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B124/B203

tration does not change under these conditions either. The calibration curve must, therefore, be determined in the presence of a certain Al amount, which permits a determination of 2-50% of Nb in an aluminum alloy without aluminum separation. While in the niobium separation by acid hydrolysis aluminum delays the hydrolytic precipitation of niobic acid, and no precipitate is formed at a niobium content below 5%, niobium always remains in the basic solution in the aluminum separation by lye, and too low results are obtained. There are 4 figures, 7 tables, and 4 Soviet-bloc references.

Table 1: Determination of niobium with Arsenazo in the presence of tartaric acid.
Legend: A) Nb content, mg, B) Nb obtained, mg, C) tartaric acid content, mg.

А) Содержание Nb, мг	Б) Получено Nb, мг	В) Содержание винной кислоты, мг
0.04	0.045	2
0.10	0.11	4
0.18	0.19	7
0.30	0.32	12
0.40	0.41	15
0.60	0.58	24
0.80	0.81	32

Card 3/6

MIKHAEVA, L.A.; NIKITINA, Ye.I.

Biotite rocks in various intrusive complexes of the Gorny Altai.
Geol. i geofiz. 10:27-36 '60. (MIA 14:2)

1. Institut geologii i geofiziki Sibirskogo otdeleniya AN SSSR,
Novosibirsk.

(Altai Mountains--Biotite)

TYCHINSKIY, A.A.; SOTNIKOV, V.I.; NIKITINA, Ye.I.

Manifestation of a new type of copper mineralization in the
southeastern Altai. Geol.i geofiz. no.12.70-79 '61. (Altai 15:5)

1. Institut geologii i geofiziki Sibirskogo otdeleniya AN SSSR,
Novosibirsk.

(Altai Mountains--Copper ores)

SOTNIKOV, V.I.; NIKITINA, Ye.I.

Behavior of accessory minerals and accessory elements in the process of greisenization. Geol. i geofiz. no.10:58-78 '63. (MIRA 17:1)

1. Institut geologii i geofiziki Sibirskogo otdeleniya AN SSSR, Novosibirsk.

NIKITINA, Ye.I.; HERZINA, A.P.; KUZNETSOVA, I.K.; SOTNIKOV, V.I.

Svanbergite in the Gornyy Altai. Dokl. AN SSSR 149 no.4:942-944
Ap '63. (MIRA 16:3)

1. Institut geologii i geofiziki Sibirskogo otdeleniya AN SSSR.
Predstavleno akademikom V.S.Sobolevym.
(Altai Mountains--Svanbergite)

SKURIDIN, V.A.; NIKITINA, Ye.I.

Geochemical characteristics of the biotites of intrusive and metamorphic formations in the Kuray Range (Gornyy Altai).
Geol. i geofiz. no.6:158-163. '64. (MIRA 18:11)

1. Institut geologii i geofiziki Sibirskogo otdeleniya
AN SSSR, Novosibirsk.

STOLETOV, V.N.; BUDNITSKAYA, Ye.V.; AGAMALOVA, S.B.; KONDASHOVA, T.A.,
NIKITINA, Ye.I.

Characteristics of the changes in nucleic acid metabolism in
ontogeny of various wheat forms. Izv. AN SSSR. Ser. biol. no. 6:
836-847 N-D '65. (MIRA 12:11)

1. Gosudarstvennyy universitet im. M.V. Lomonosova. Institut
biokhimii im. A.N. Bakha AN SSSR.

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LITVINOV, N.S., inzh.; NIKITINA, Ye.L., inzhener-khimik;
GROKHOL'SKAYA, S.D., laborant

Method of determining the turbidity of water by means of
the FEK-N-57 apparatus. Gig. i san. 28 no.7:48-49 J1 '63.
(MIRA 17:1)

1. Iz sanitarno-bakteriologicheskoy laboratorii desnyanskogo
vodoprovoda Kiyeva.

LITVINOV, N.S.; NIKITINA, Ye.L.; GROKHOL'SKAYA, S.D.

Determination of the color index of drinking water with the
FEK-N-57 apparatus. Gig. i san. 28 no.6:54-55 Je'63 (MIRA 17:4)

1. Iz laboratorii desnyanskogo vodoprovoda Kiyeva.

Nikitina, E. N.

USSR/Chemistry - Catalytic conversion

Card 1/1 Pub. 151 - 22/38

Authors : Gavrilov, B. G., and Nikitina, E. N.

Title : Thermocatalytic conversions of butylnaphthalene

Periodical : Zhur. ob. khim. 24/2, 303-307, Feb 1954

Abstract : Thermocatalytic conversion of mono- and di-secondary-butylnaphthalins over a natural aluminum silicate catalyst was investigated. In addition to the reactions leading to the displacement of the immutable fatty radicals, which are typical for alkylbenzenes, numerous other reactions were also observed. The most characteristic of these reactions were the formation of diethylbenzene, tetrahydronaphthalin, dinaphthyl and butane which take place through the over-distribution of hydrogen, and the formation of octane (3,4-dimethylhexane) due to the combination of butyl radicals. The results obtained confirm the general law regarding the processes of petroleum conversion in nature: aromatic hydrocarbons → naphthene hydrocarbons → methane hydrocarbons. Nine references: 1-English and 8-USSR (1928-1953). Tables.

Institution : The A. A. Zhdanov State University, Leningrad

Submitted : September 5, 1953

Ref Zhur, No. 3, 1957, No. 6611

Category : USSR/Solid State Physics - Phase Transformation in Solid Bodies E-5

Abs Jour : Ref Zhur - Fizika, No. 3, 1957, No. 6611

Author : Kalobnev, I. F., Aristova, N. A., Bernshteyn, M. L., Nekitina, Ye.N.

Title : Use of the Ultraviolet Microscope in the Investigation of the Structure of Aluminum Alloys.

Orig Pub : Zaovd, laboratoriya, 1956, 22, No. 7, 803-804

Abstract : No abstract

NIKITINA, YE. N.

- Abstracts from USSR. *Kachestvo po analiticheskoy khimii*
Metody opredeleniya prirody i chistoty metallov (Methods of Determining Abi-
 tures in Pure Metals) Moscow, 1960. 111 p. (Series: It. 77-47, 12) 3,500
 copies printed.
- See: 144: A. P. Vinogradov, Academician, and D. I. Pechenkin, Doctor of Chemical
 Sciences; Ed. of Publishing House: M. P. Volynskii; Tech. Ed.: V. P. Volynskii.
- INDEX: This collection of articles is intended for chemists, metallurgists, and
 engineers.
- CONTENTS: The articles describe methods for detecting and determining various ad-
 mixtures and their traces in pure metals. Also discussed are many chemical,
 physicochemical, electrochemical, spectrochemical and luminescence methods of
 analyzing materials of high purity. The editors state that these methods have
 been developed within the last five or six years by various Soviet scientific
 institutes, and are now widely used in research and factory laboratories of the
 Soviet Union. No preconditions are mentioned. References, mostly Soviet,
 accompany each article.
- Aliev, R. B., P. B. Galimov, K. A. Subhanov, and O. B. Fakhimov. Determina-
 tion of the Oxygen and Hydrogen Content in Solid Samples of Niobium
 and Zirconium by the Spectral Method 289
- Bobkov, I. B., A. A. Pishchikov, and I. A. Chernyshova. Determination of
 traces of lead, tin, bismuth and cadmium in metallic chromium and in its
 alloys 298
- Blatt, Ya. B. Determination of Abundance of Antimony in Pure Chromium
 and in its Alloys 311
- Borovoy, G. A. Spectral Determination of Abundance of Bismuth, Cadmium,
 Tin, Lead and Antimony in Chromic Oxide and in Chromic Anhydride 314
- Budim, I. M., G. A. Petukov, and I. I. Voronin. Spectrochemical method
 of determining abundance of bismuth, cadmium, tin, lead, and antimony in
 chromium anhydride 317
- Chernyshova, I. A., and R. A. Pishchikov. Application of Acetylene A-C
 Spectroscopy to Determine Small Quantities of Sodium, Calcium, and
 Lithium Abundance in Metallic Niobium and Zirconium 322
- Chernyshova, I. A., R. I. Petukov, I. I. Voronin, and V. M. Litvinov. A
 Determination of Abundance in Niobium and Zirconium Oxide 331
- Chernyshova, I. A., and Z. M. Furmanova. Determination of Oxygen in Metallic
 Beryllium 341
- Chernyshova, I. A., Ya. B. Blatt, and V. V. Zhukovskaya. A. E.
 Chudakov, and I. I. Voronin. Luminescence method for the quantitative
 determination of cadmium in metallic beryllium 344
- Chudakov, I. I., and I. I. Voronin. Application of the Luminescence Method
 to Determine Small Quantities of Nickel, Bismuth, and Lead in Chromium and
 its Alloys 355
- Chudakov, I. I., and I. I. Voronin. Spectral Analysis of High-Purity Nickel
 Abundance 366
- Chudakov, I. I., and I. I. Voronin. Spectral Analysis of Small Quantities of
 Antimony, Lead, and A. A. Pishchikov. Separation of Small Quantities of
 Cobalt from Large Quantities of Nickel 377
- Chudakov, I. I., and R. A. Pishchikov. Mass Analysis of Nickel-Base Alloys 383
- Chudakov, I. I., R. A. Pishchikov, and I. I. Voronin. Determination of Small
 Quantities of Cadmium, Strontium, and Barium in Metallic Chromium 393

ANALYST: Library of Congress

10/11/11, Ye. S.

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[illegible]

NIKITINA, Ye. S.

USSR/Metals-Plating
Chemistry-Analysis, Metal

Jun 50

"Rapid Determination of Some Metals in Coating on Iron," A. I. Kogan, Ye. S. Nikitina, Odessa Electrotech Inst of Communications

"Zavod Lab" Vol XVI, No 6, pp 672-674

Suggests 40% solution of mercuric nitrate as reagent for removing zinc, tin, lead, copper, and their alloys from surface of iron in process of determining thickness of coatings. Reagent has almost no effect on iron. Gives results of determination, by this method, of zinc in galvanized iron and tin in tin plate.

PA 163T57

AUTHORS: P'yankov, V.A., Nikitina, Ye.S., Kostyuk, A.P. *JN/78-3-7-24/44*

TITLE: On the Interaction Between Zinc and Oxygen in Solutions of Alkaline Halides (O vzaimodeystvii tsinka s kislorodom v rastvore galogenidov shchelochnykh metallov)

PERIODICAL: Zhurnal neorganicheskoy khimii. 1958, Vol. 3, Nr 7, pp 1608-1610 (USSR)

ABSTRACT: The velocity of the reaction of zinc with oxygen in solutions of chlorides, bromides, and iodides of potassium at various temperatures and various concentrations of the reacting substances was investigated. The reaction velocity of the interaction between zinc and oxygen increases from iodide to chloride. The reaction develops probably according to the following scheme:

$$2 \text{Zn} + \text{O}_2 + 8 \text{Cl}^- + 2 \text{H}_2\text{O} = 2 \text{ZnCl}_4^{2-} + 4 \text{OH}^-$$
 The results indicate that in the first stage of this reaction unstable zinc-halide complex salts are formed from the solutions of which the surplus zinc-portion is precipitated while zinc hydroxide or basic zinc halide is formed. There is a linear connection between the concentration of oxygen and the quantity of zinc. The

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On the Interaction Between Zinc and Oxygen in Solutions
of Alkaline Halides

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concentration of the halides exercises comparatively little influence upon reaction velocity. With an increase of halide concentration to 16 times its amount, reaction velocity increases by 3 to 4 times its amount. Also the concentration of zinc in the solution exercises only little influence on the velocity of reaction. There are 3 figures, 3 tables and 4 references, 3 of which are Soviet.

SUBMITTED: June 28, 1957

1. Zinc--Chemical reactions
2. Oxygen--Chemical reactions
3. Alkali halide solutions--Chemical properties

Card 2/2

NIKITINA, Yekaterina Trofimovna; SARTBAYEVA, Uriya Abdukalykovna;
BALITSKAYA, A.K., kand. veter. nauk, otv. red.;
RZHONDKOVSKAYA, L.S., red.; KHUDYAKOV, A.G., tekhn.red.

[Microbial antagonism and antibiotics] Antagonizm mikrobov
i antibiotiki. Alma-Ata, Izd-vo AN KazSSR, 1963. 39 p.
(MIRA 17:1)

NIKITINA, Ye.T.; ALEKSEYEVA, Z.I.

Antibiotic properties of the group of blue and violet actinomyces from the soils of Kazakhstan. Trudy Inst. mikrobiol. i virus. AM Kazakh. SSR 7:147-156 '63 (MIRA 14:12)

NIKITINA, Ye. T.: Master Biol Sci (diss) -- "Bacteria which are antagonistic to *Fusarium oxysporum* var. *solani*, and possibilities for their practical utilization". Alma-Ata, 1952. 150 pp (Kazakh State Univ S. M. Kirov, Coll Biology Faculty), 150 copies (KL, No 4, 1952, 124)

ORIGIN
DATE

U.S.S.R.
Microbiology

APR. 1959

Ref Zhur-Biolosiya, No.4, 1959, No. 14774

AUTHOR
INST.

Nikitina, Ye.T.

TITLE

Microbiotic bacteria and their role in the
protection of plants from Fusarium wilt

ORIG. PUB.

Tr. In-ta mikrobiol. i virusol. AN Kaz3SR,
1958, 2, 24-41

ABSTRACT

Microbiotic bacteria - antagonistic to Fusari-
um oxysporum var. solani (agent of early wilt
in the potato) - were isolated from the soil
of Alma-Atinskaya Oblast'. The most active
strain was isolated from the rhizosphere of
alfalfa, clover, and esparto. A detailed
study was made of 5 strains of bacteria which
produced thermolabile, microbatic, and micro-
static factors. The high relative humidity
(90 - 100%) and high temperature (30 - 37 de-

CARD:

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The least number was found when potatoes were
planted after alfalfa (1.2 thousand in 1 m²)

COUNTRY : USSR 0
 CATEGORY : Plant Diseases. Cultivated Plants.
 ABS. JOUR. : RZhBiol., No. 3, 1959, No. 11288
 AUTHOR : Nikitina, Ye. T.
 INST. : Institute of Microbiology and Virusology, AS Kazakh SSR
 TITLE : On the Problem of the Etiology of the Premature Wilt of Potatoes.
 ORIG. PUB. : Tr. In-ta mikrobiol. i virusol. AN KazSSR, 1958, 2, 66-79
 ABSTRACT : Of the potato plants affected with the premature wilt, the fungi of the genus *Fusarium* were isolated in 32% of the cases. In 98% of the cases, there is observed in these plants damage of the root system and of the above-ground part of the stem (unlike the physiological and the big bud wilt). This furnishes a reason for considering the fungi of the genus *Fusarium* to be the causative agent of the infectious potato wilt under the conditions of Alma-Ata suburban zone. For the control of this disease, measures are recommended which improve the air conditions.

CARD: 1/2

-9-

NIKITINA, Ye.T.

Role of alfalfa in ridding soils of the pathogen producing the
fusarium wilt of potatoes. Trudy Inst. Mikrobiol. i virus. AN
Kazakh. SSR 3:139-147 '59. (MIRA 13:2)
(ALFALFA) (POTATOES--DISEASES AND PESTS)
(FUNGI, PHYTOPATHOGENIC)

NIKITINA, Ye.T.; LAPUKHINA, G.P.

Causative agent of black bacterial mottling in tomatoes on the farms
of the Alma-Ata suburban zone. Trudy Inst. mikrobiol. i virus. AN
Kazakh. SSR 4:140-145 '61. (MIRA 14:4)
(BACTERIA, PHYTOPATHOGENIC) (TOMATOES—DISEASES AND PESTS)

NIKITINA, Ye.T.; ISABAYEVA, M.K.

Antibiotic activity of the fungi of the genus Trichoderma found
in the soils of Kazakhstan. Trudy Inst. mikrobiol. i virus.
AN Kazakh. SSR 5:39-43 '61. (MIRA 15:4)
(Kazakhstan--Trichoderma) (Antibiotics)

SEYKETOV, G.Sh.; NIKITINA, Ye.T.

Parasitic characteristics of fungi of the genus Trichoderma,
isolated from the soils of Kazakhstan. Trudy Inst.mikrobiol.i
virus.AN Kazkah.SSR 6:42-47 '62. (MIRA 15:8)
(KAZAKHSTAN—TRICHODERMA)

NIKITINA, Ye.T.; ISABAYEVA, M.K.; AMIRKHANOVA, L.

Volatile antibiotics from four fungus species of the genus Trichoderma. Trudy Inst.mikrobiol.1 virus.AN Kazkah.SSR 6:48-52 '62.
(MIRA 15:8)

(TRICHODERMA) (ANTIBIOTICS)

NIKITINA, Ye.T.; LEVINA, A.A.; ISABAYEVA, M.K.

Specific composition and antibiotic characteristics of the genus
Trichoderma in various soil types of Kazakhstan. Trudy Inst.
mikrobiol. i virus. AN Kazkah. SSR 6:53-60 '62. (MIRA 15:8)
(KAZAKHSTAN—TRICHODERMA) (ANTIBIOTICS)

NIKITINA, Ye.T.

Use of antibiotics and other microbial metabolites in animal
husbandry. Izv. AN Kazakh. SSR. Ser. biol. nauk 2 no.3:99-102
My-Je '64. (MIRA 17:10)

NIKITINA, Ye.T.; ALEKSEYEVA, Z.I.

Taxonomic position of the group of actinomycete antagonists.
Trudy Inst. mikrobiol. i virus. AN Kazakh. SSR. 8:42-64, '65.

Taxonomic position of the group of violet actinomycete
antagonists. Ibid.:65-74 (MIRA 18:11)

BELYAZO, Ivan Afanas'yevich, DMITRIYEV, Valeriy Razumnikovich, ~~NIKITINA~~
Yelena Vasil'yevna, PESTRIKOV, Aleksandr Nikolayevich, ~~BAKITO~~, Z.I.
-red.

[Electric interlocking systems] Elektricheskie releinye tsentralizatsii.
Moskva, Gos. transp. shel-dor. izd-vo, 1958. 195 p. (MIRA 11:9)
(Railroads--Signalling--Interlocking systems)

BELYAZO, Ivan Afanas'yevich; DMITRIYEV, Valeriy Razumnikovich; NIKITINA, Yelena Vasil'yevna; PESTRIKOV, Aleksandr Nikolayevich; ZHIL'TSOV, P.N., inzh., retsenzent; MARENKOVA, G.I., inzh., red.; MEDVEDEVA, M.A., tekhn. red.

[Route-relay interlocking systems] Marshrutno-releinaia tsentralizatsiia. Moskva, Vses. izdatel'sko-poligr. ob"edinenie M-va putei soobshcheniia, 1962. 282 p. (MIRA 15:5)

(Railroads--Signaling --Block system)
(Railroads--Signaling--Interlocking systems)

PHASE I BOOK EXPLOITATION

1083

Belyazo, Ivan Afanas'yevich, Dmitriyev, Valeriy Razumnikovich,
Nikitina, Yelena Vasil'yevna, and Pestrikov, Aleksandr Nikolayevich

Elektricheskiye releynyye tsentralizatsii (Electric Interlocking
Systems) Moscow, Transzheldorizdat, 1958. 195 p. 5,000 copies
printed.

Ed.: Rakito, E. I.

PURPOSE: This monograph is addressed to engineering and technical
workers employed in railroad signalling and communications.

COVERAGE: The book discusses standardized circuits of centralized
traffic control systems, which are used today (regardless of the
system of control) in designing and constructing electric relay
interlocking systems. The function of circuit components and the
operation of the circuits as a whole are described for interlock-
ing systems with sectional control. The book describes plug re-
lay designs and presents reference material on relays and trans-

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Electric Interlocking Systems 1083

formers. There is an insert containing connection diagrams of the interlocking relays discussed in the text. Giprotrans-signalsvyaz' (State Institute for the Design of Railroad Signaling and Communications Equipment) is credited with having developed in 1945 and 1946 two interlocking systems. These systems are described in the present work. No personalities are mentioned. There are no references.

TABLE OF CONTENTS:

Ch. 1. General Principles of Relay Interlocking	3
1. Basic aspects	3
2. Components of interlocking equipment	5
Ch. II. Control Equipment	6
1. Track indicator control panel	6
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3. Dwarf signal post	16

Card 2/7

NIKITINA, Ye.V.

Studying aluminosilicate catalysts obtained from impregnation
of silica gel by aluminum nitrate. Izv. vys. ucheb. zav.; neft' i
gaz 2 no.7:69-73 '59. (MIRA 12:12)

1. Groznenskiy neftyanoy institut.
(Aluminum silicates)

NIKITINA, Ye.V.; OBORIN, V.I. [deceased]

Comparative study of methods for obtaining aluminosilicate catalysts.
Izv. vys. ucheb. zav.; neft' i gaz 3 no.10:75-81 '60. (MIRA 14:4)

1. Groznenskiy neftyanoy institut.
(Aluminosilicates)

S/152/63/000/002/003/003
B126/B186

AUTHORS: Panchenkov, G. M., Nikitina, E. V.

TITLE: Exchangeability of aluminosilicates prepared by various methods

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Neft' i gaz, no. 2,
1963, 70 - 74

TEXT: Three methods were used to prepare samples of an aluminosilicate catalyst: blending of sulfate of aluminum solutions and liquid glass with subsequent coagulation to aluminosilicate gel; saturation of humid silica gel with solutions of aluminum nitrate of various concentrations with subsequent decomposition of aluminum salt during drying and heating; and blending of separately precipitated silica gels and aluminum hydroxide. It was shown that the exchange of aluminum and hydrogen ions in aluminosilicates obtained by joint precipitation is subject to the law of mass, and their maximum quantity exchangeable against the ion of sodium from a sodium chloride solution was calculated on this basis. However, this law is not applicable to samples obtained by saturation of silica gel or by blending separately precipitated gels. In these catalysts the quantity of exchangeable ions of aluminum is much smaller. However, an increase
Card 1/2

NIKITINA, Ye. V.

Vegetation of mountain winter pastures in the western coastal
region of Issyk-Kul. Trudy Biol. inst. Kir PAN SSSR no. 1:7-39
'47.

(MLRA 8:10)

(Issyk-Kul region--Pastures and meadows)

1. NIKITINA, YE. V.
2. USSR (600)
4. Plants - Tien Shan
7. Influence of grazing on the changing plant complex of the Tien Shan range. Izv. KirFAN SSSR no. 7, 1947.

9. Monthly List of Russian Accessions, Library of Congress, January 1953. Unclassified.

NIKITINA, Ye.V.

Modifications in the wormwood-feather grass semidesert under irrigation. Trudy Biol.inst.KirFAN SSSR no.3:103-106 '50. (MIRA 8:5)
(IRRIGATION)
(DESERT FLORA)

NIKITINA, Ye.V.

Timely utilization of pastures as a method of controlling plant
poisoning among animals. Trudy Biol.inst.KirFAN SSSR no.3:107-111
'50.

(MLRA 8:5)

(PASTURES)

(POISONOUS PLANTS)

NIKITINA, Ye.V.

Monocotyledonous plants in Kirghizia. Trudy Biol. inst. KirFAN SSSR
no.4:175-178 '51.

(MLRA 9:10)

(KIRGHIZISTAN--MONOCOTYLEDONS)

NIKITINA, YE.V.

The Committee on Stalin Prizes (of the Council of Ministers USSR) in the fields of science and inventions announces that the following scientific works, popular scientific books, and textbooks have been submitted for competition for Stalin Prizes for the years 1952 and 1953. (Sovetskaya Kultura, Moscow, No. 22-40, 20 Feb - 3 Apr 1954)

<u>Name</u>	<u>Title of Work</u>	<u>Nominated by</u>
Nikitina, Ye.V.	"Flora of the Kirgiz SSR"	Kirgiz Affiliate of the
Rozhenits, R.Yu.		Academy of Sciences USSR
Kashchenko, L.I.		
Protopopov, G.D.		
Popova, L.I.		
Shishkin, B.K.		
Vvedenskiy, A.I.		

SO: W-30604, 7 July 1954

NIKITINA, Ya. V.

Formation of the Botanical garden of the city of Frunze. Trudy Inst.
bot. i rast. KirWAN SSSR no.1:45-48 '54. (MLRA 10:1)
(Frunze--Botanical gardens)

NIKITINA, Ye.V.; PROTOPOPOV, G.F.; ROZHEVITS, R.Yu. [deceased]; POPOVA, K.I.,
 LASHCHENKO, L.I.; SMIRNOV, L.A.; TKACHENKO, V.I.; YAKUBOVA, P.A.;
 GOLOVKOVA, A.G.; AYDAROVA, P.A.; SHPOTA, Ye.I.; SHEVCHENKO, D.A.;
 SHISHKIN, Boris Konstantinovich, professor, doktor biologicheskikh
 nauk, nauchnyy redaktor; VVEDENSKIY, A.I., nauchnyy redaktor;
 YEVVRUSHENKO, G.A., professor, otvetstvennyy redaktor; KOVALEV, V.N.,
 otvetstvennyy redaktor; SEREBRYAKOV, V.I., tekhnicheskii redaktor

[The flora of Kirghizistan; classification of the plants of
 Kirghizistan] Flora Kirgizskoi SSR; opredelitel' rastenii Kirgizskoi
 SSR. Sost. E.V.Nikitina i dr. Frunze, Izd-vo Akademii nauk Kirgizskoi
 SSR. Vol.1. [Pteridophyta, Gymnosperms and Monocotyledons of the
 Angiosperms] Paprotnikoobraznye, golosemennye i odnodol'nye iz
 pokrytosemennyykh. 1952. 103 p. Vol. 2. [Grasses and sedges] Zlaki
 i osokovyie. 1950. 315 p. Vol.3. [Aroidae - Orchidaceae] Aroidnye -
 Orkhidnye. 1951. 148 p. Vol.4. [Salicaceae - Polygonaceae] Ivovye -
 Grechishnye. 1953. 153 p. Vol. 5. [Families: Chenopodiaceae,
 Amaranthaceae, Portulacaceae, Caryophyllaceae] Semeistva: Marevyie,
 Amarantovyie, Portulakovyie, Gvozdnichnye. 1955. 185 p. Vol. 6.
 [Families: Ceratophyllaceae, Ranunculaceae, Berberidaceae,
 Papaveraceae, Capparidaceae, Cruciferae] Semeistva: Rogolistnikovyie,
 Liutikovye, Barbarisovyie, Makovyie, Kapersovyie, Krestotsvetnye. 1955.
 297 p. (MLRA 9:10)

1. Chlen-korrespondent Akademii nauk SSSR (for Shishkin)
 (Kirghizistan--Botany)

POSPELOV, A.G.; CHAGOROVA, R.M.; NIKITINA, Ye.V., redaktor; TSYBINA, Ye.V.,
tekhnicheskii redaktor

[Wood destroying house fungi and measures to control them;
a popular scientific brochure.] Derevorazrushaiushchie
domovye griby i mery bor'by s nimi; nauchno-populiarnaiia
broshiura. Frunze, Izd-vo Akad. nauk Kirgizskoi SSR, 1957.
28 p.

(Wood-decaying fungi)

(MLRA 10:5)

ADD LIGHT, C/R v
GAREYEV, E.Z., kand.sel'skokhoz.nauk; TKACHENKO, V.I., kand.biolog.nauk;
KUNCHENKO, A.I., mladshiy nauchnyy sotr.; SHPAK, R.L., mladshiy
nauchnyy sotr.; KRIVOSHEYEVA, L.S., mladshiy nauchnyy sotr.;
NIKITINA, Ye.V., kand.biolog.nauk, red.; ANOKHINA, M.G., tekhn.red.

[Guide to the botanical garden] Putevoditel' po Botanicheskomu
sadu. Frunze, 1957. 78 p. (MIRA 11:1)

1. Akademiya nauk Kirgizskoy SSR, Frunze. Botanicheskiy sad.
2. Akademiya nauk Kirgizskoy SSR, Botanicheskiy sad, Institut
botaniki (for Kareyev, Tkachenko, Kunchenko, Shpak, Krivosheeva,
Nikitina).

(Frunze--Botanical gardens)

MIRNAYA V.
POSPELOV, A.G.; ZAPROMETOV, N.G.; DOMASHEVA, A.A.; NIKITINA, Ye.V., red.;
TSYBINA, Ye.V., tekhn.red.

[Fungi of the Kirghiz S.S.R.] Gribnaya flora Kirgizskoi SSR.
Frunze, Izd-vo AN Kirgizskoi SSR. No.1. [Systematic list of
species and geographical distribution] Sistematischesko-vidovoi
sostav i geograficheskoe rasprostranenie. 1957. 128 p.
(Kirgizistan--Fungi) (MIRA 11:6)

NIKITINA, Ye.V.; POPOVA, L.I.; AYDAROVA, R.A.; KASHCHENKO, L.I.; PROTOPOPOV,
G.F.; UBUKEYEVA, A.U.; TKACHENKO, V.I.; KORNEVA, I.G.; OBOZOV, A.O.;
GOLOVKOVA, A.G.; VVEDENSKIY, A.I., nauchnyy redaktor; TSYBINA, Ye.V.,
tekhnicheskiiy redaktor

[Flora of the Kirghiz S.S.R.; guide to plants of the Kirghiz S.S.R.]
Flora Kirgizskoi SSR; opredelitel' rastenii Kirgizskoi SSR. Frunze,
Izd-vo AN Kirgizskoi SSR. Vol.7. 1957. 642 p. (MLRA 10:9)
(Kirghizistan--Botany)

USSR/General Division. Conservation of Nature.

A-5

Abs Jour: Ref. Zhur. Biologiya, No 4, 1958, 14243.

Author : Nikitina E.V., Yamushevich A.I.

Inst :

Title : The Conservation of Nature in Kirgizia

Orig Pub: Ukhrana prirody i zapovedn. delo v SSSR, 1957, No 2, 16-24

Abstract: The immediate measures for the conservation of nature in Kirgizia are: 1) organization of reserves: Fergana (walnut forests and thickets of pistachio), Talas (fir-tree forests and juniper), Altyn Arasan (fir forests), and others; 2) the putting-in-order of the forest and hunting economy; 3) the discontinuation of burning thickets of "oblepikhi"; [a tall thorned bush with yellow sourish edible berries] 4) the adjustment of the conservation of water fowl and pheasants in the area of Lake Issyk-Kul.

Card : 1/1

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NIKITINA, Ye.V.; DZHANAYEVA, V.M., red.; ANOKHINA, M.G., tekhn.red.

[Poisonous, roxious, and inedible plants in pastures of the
Kirghiz S.S.R.] Iadovitye, vrednye i nepoedaemye rasteniia
pastbishch Kirgizskoi SSR. Frunze, Akad.nauk Kirgizskoi SSR,
1959. 55 p. (MIRA 13:7)
(Kirghizistan--Pastures and meadows) (Weeds)